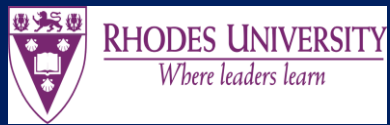


CHARMM force field parameters for the Zn²⁺ centre of 6-pyruvoyl tetrahydropterin synthase enzyme

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Introduction

The malaria parasite drug-resistance has raised a great challenge in anti-malarial drug discovery and grounded the need for new treatments. 6-pyruvoyl tetrahydropterin synthase (PTPS) is the second enzyme of the malaria parasite *de novo* folate biosynthesis pathway. PTPS is responsible for the conversion of dihydroneopterin triphosphate to 6 pyruvoyl tetrahydropterin via a base catalyzed redox transfer reaction and elimination of the triphosphate tail [3]. The hexameric enzyme 3D structure comprises two symmetrical trimers, each has three monomers. The enzyme has six zinc-containing active sites, each buried in a deep pocket of 12 Å [3]. The active site Zn²⁺ ion is coordinated to three histidine residues through their NE2 atoms (Figure 2). In this study, appropriate force fields parameters describing the Zn²⁺ ion were developed, using quantum mechanics (QM) calculations and then validated through all atomics molecular dynamics (MD) simulations. The generated parameters are of important use for accurate MD simulations for future computer-aided drug discovery studies.

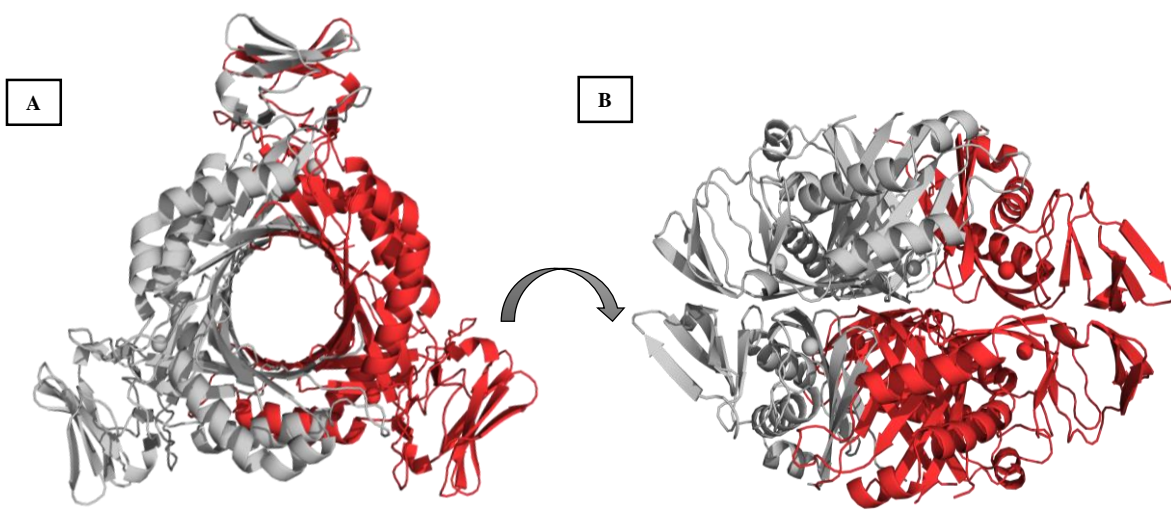


Figure 1: (A) top view and (B) side view of *Plasmodium falciparum* PTPS homo-hexameric enzyme structure (PDB ID: 1Y13).

Methodology

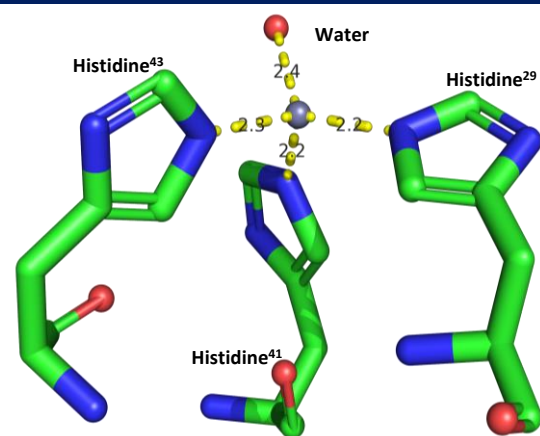
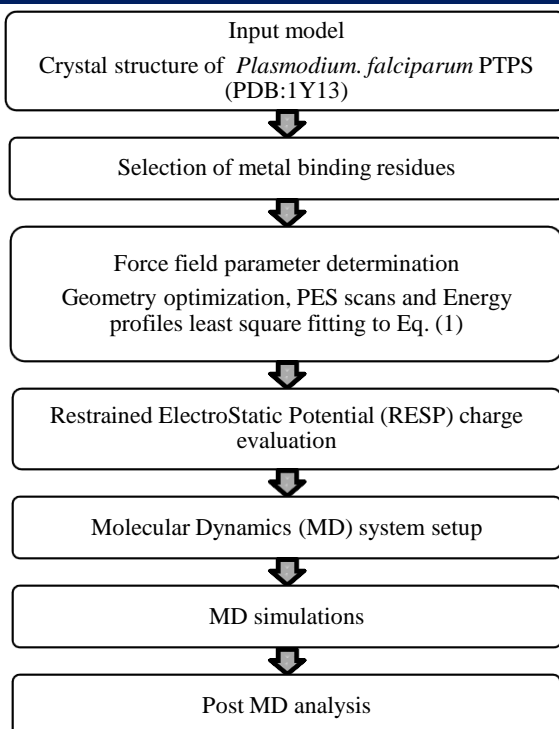


Figure 2: Active site subset selected from the x-ray structure (PDB ID: 1Y13).

$$E_{total} = \sum_{bonds} k_b(r - r_{0,b})^2 + \sum_{angles} k_a(\theta - \theta_{0,a})^2 + \sum_{dihedrals} k_{d,n}[1 + \cos(n\chi - \delta_{d,n})] + \sum_{non-bonded} \epsilon \left[\left(\frac{R_{min,ij}}{r_{ij}} \right)^{12} - \left(\frac{R_{min,ij}}{r_{ij}} \right)^6 \right] + \frac{q_i q_j}{\epsilon r_{ij}}$$

Equation 1: CHARMM energy function

Results and Discussion

Geometry optimization

PTPS initial subset structure geometry was obtained from the crystal structure and subjected to optimization via Gaussian 09 software [4] using the density functional theory (DFT) with the Becke three-parameter hybrid exchange functional and the Lee-Yang-Parr (B3LYP) correlation functional [5] [6] [7]. The LanL2DZ pseudopotential and associated basis functions were used to describe the Zn²⁺ ion and the 6-31G (d) basis set was used for the organic atoms. The optimised values of bonds distance and angles were captured and compared to the initial X-ray structure (Table.1).

Table 1. The optimised bond lengths (Å), angles (°) compared to the initial X-ray structure.

	Overlay similarity	Bond lengths (Å)			Angles (°)		
		Zn ²⁺ -(NE2) HIS ²⁹	Zn ²⁺ -(NE2) HIS ⁴¹	Zn ²⁺ -(NE2) HIS ²⁹	HIS ⁴¹ NE2 - Zn ²⁺ - HIS ²⁹ NE2	HIS ⁴¹ NE2- Zn ²⁺ - HIS ⁴³ NE2	HIS ²⁹ NE2- Zn ²⁺ - HIS ⁴³ NE2
Crystal structure 1Y13	1	2.29	2.13	2.25	91.22	94.43	97.55
QM (B3LYP/6-31G*)	0.71	2.03	2.03	2.04	115.26	114.14	117.96
MD		2.18	2.17	2.16	81.65	80.94	86.45

RESP charge evaluation

RESP Charges calculation was performed at the DFT/B3LYP level of theory using 6-31G (d) basis set for the organic atoms [5] [6] and LanL2DZ pseudopotential and associated basis functions [7] for the Zn²⁺. The evaluated RESP potential for the Zn²⁺ was identified to be +0.80, the rest of the residues charges and their corresponding atom types are shown in Figure 3.

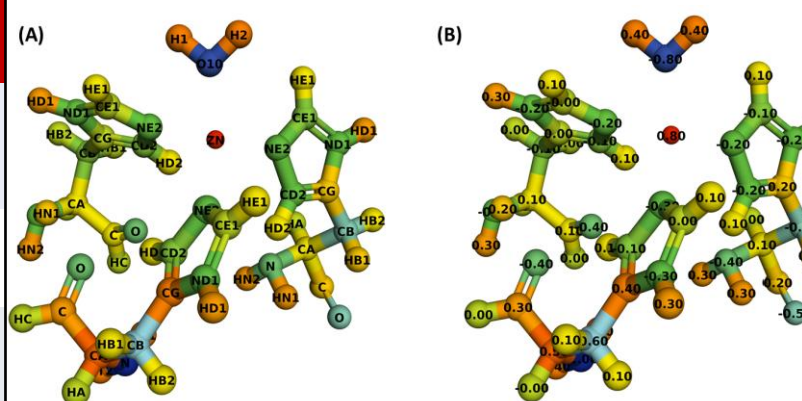


Figure 3: PTPS active site subset. (A) Atom types and (B) RESP charges

Force field parameters

The resources available at the Centre for High Performance Computing (Cape Town, South Africa) were used to accommodate the computational cost and time associated with performing such calculations. The QM calculations resulted in the generation of energy profiles for the bonds stretching, angles bending and dihedrals rotation. The energy profiles exhibited a harmonic profile and were reasonably following the trend of the theoretical (MM) data (Figure 4). Table 2 summarizes the derived force field parameters.

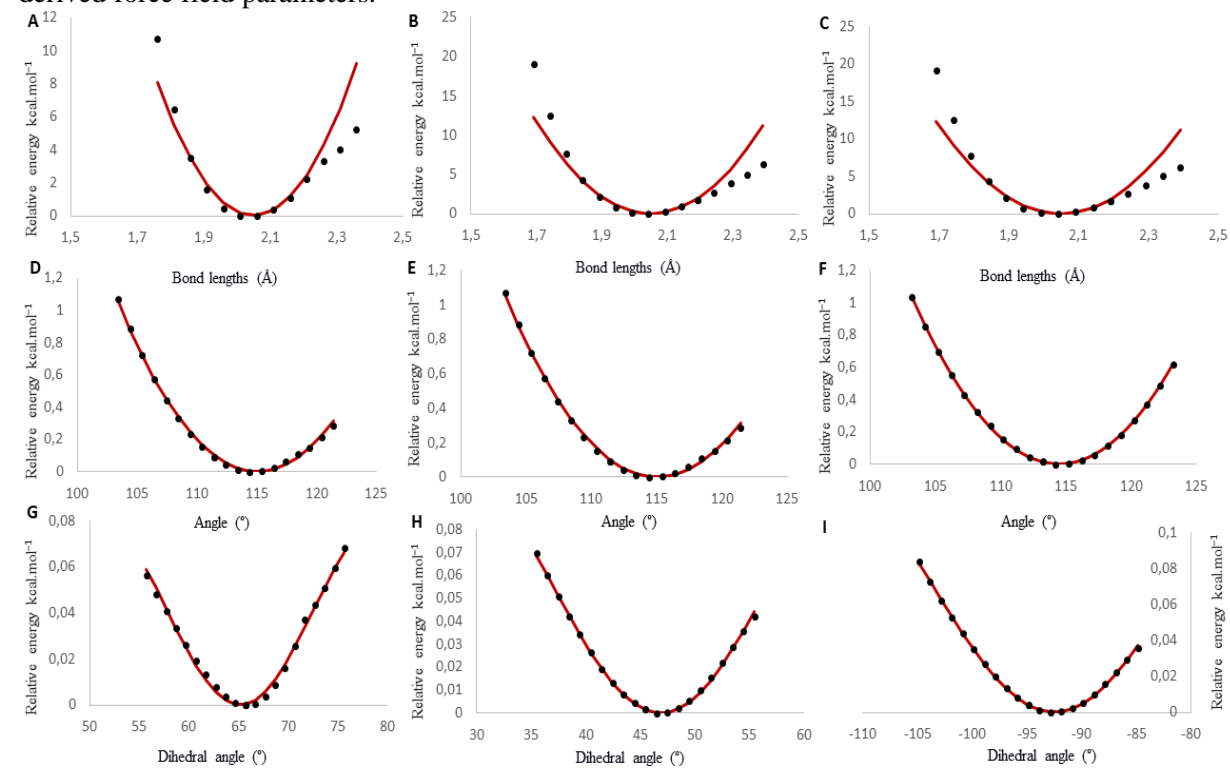


Figure 4: Potential energy surface scans of the Zn²⁺ coordinating residues of the PTPS protein. Forward and reverse PES scans were performed for three bonds (A-C), three angles (D-F) and two dihedral angles (G and H).

Table 2: Developed force field parameters of the PTPS enzyme active site.

Bonds	Kr (kcal mol ⁻¹ Å ⁻²)	r _{eq} (Å)	
Zn ²⁺ -NE2 ⁴¹	95.86	2.01	
Zn ²⁺ -NE2 ²⁹	95.86	2.01	
Zn ²⁺ -NE2 ⁴³	95.86	2.09	
Angles	K _θ (kcal mol ⁻¹ rad ⁻²)	θ _{eq} (degrees)	
NE2 ²⁹ -Zn ²⁺ -NE2 ⁴³	25.53	115.04	
NE2 ²⁹ -Zn ²⁺ -NE2 ⁴¹	25.53	115.04	
NE2 ⁴¹ -Zn ²⁺ -NE2 ⁴³	26.69	114.46	
Dihedral	V _n (kcal mol ⁻¹)	n	γ
NE2 ⁴³ -Zn ²⁺ -NE2 ²⁹ -CD2 ²⁹	0.08	12.41	-14.16
NE2 ⁴³ -Zn ²⁺ -NE2 ⁴¹ -CD2 ⁴¹	0.10	9.16	-14.50
NE2 ²⁹ -Zn ²⁺ -NE2 ⁴¹ -CD2 ⁴¹	0.14	8.17	-3.77

MD simulations

MD simulations were performed using the Chemistry at Harvard Macromolecular Mechanics (CHARMM) molecular dynamics simulation and analysis computer software package [9]. The MD trajectories of 20 ns were analysed to establish the protein stability and to deduce the validity of the integrated force field parameters (Figure 5.b). The mean distance to the Zn²⁺ from the three coordinating residues was captured and shown to be maintained throughout the simulations (Figure. 5.a).

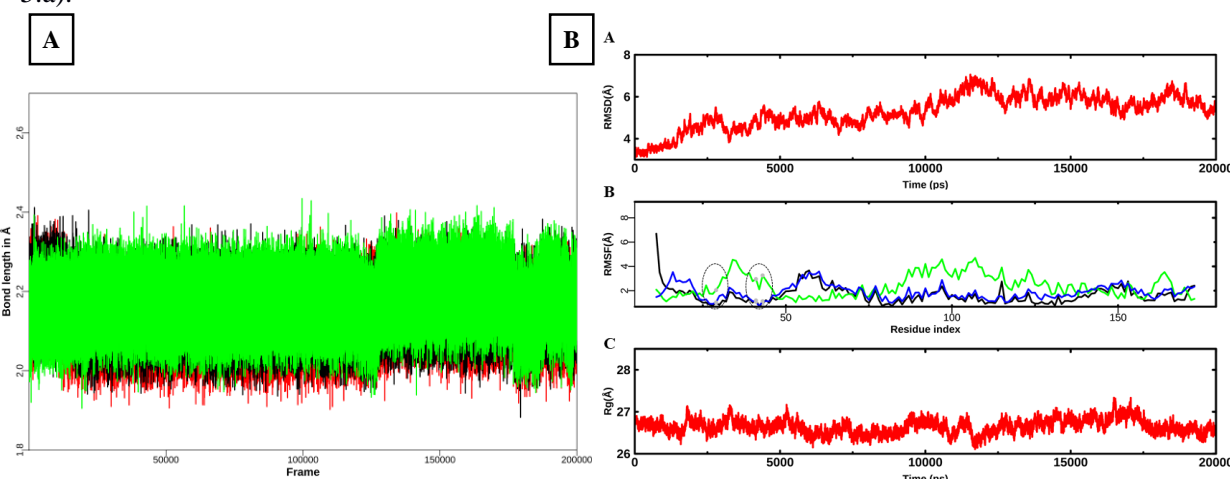


Figure.5 A: Coordination bond distance during the MD simulations. B: Stability of the PTPS protein as determined by B1: Root Mean Square Deviation (RMSD), B2: Root Mean Square Fluctuation and B3: Radius of gyration (Rg).

Conclusion

In this study, QM calculations and PES scans were performed using DFT to generate Zn²⁺ force field parameters suitable for classical MD simulations. The PES scans and the RESP atomic charge calculations were both performed at the DFT/B3LYP level of theory. The generated parameters were validated by performing a short MD simulations. The metal was shown to be stable during the simulations, suggesting that the newly generated parameters were adequately describing the coordination environment of the Zn²⁺ in the protein active site. The use of HPC was essential in obtaining these results. It has accelerated and opened new perspectives to this study, by allowing the use of parallel processor networks to study our large biomolecule system.

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